

Two-dimensional Modelling of a Non-isothermal PROX Reactor with Water Cooling for Fuel Cell Applications

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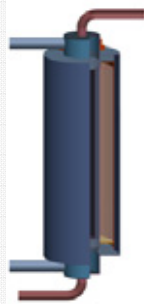
1. Introduction

- Modelling of a PROX reactor using COMSOL Multiphysics on the basis of reaction kinetics from experimental investigations

- Hydrocarbon reformat purification
- Feed gas into PROX contains 0.5 mole-% CO
- Product target CO mole fraction down to a few ppm
- Exothermic reactions
- Accurate cooling conception is mandatory

- Reactor design

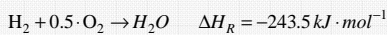
- Annular-gap-shaped PROX reactor
- Inner and outer, coaxial cooling
- Fuel cell cooling water used to cool PROX reactor and feed gas
- Cooling water temperature and the educt gas temperature are assumed to be equal



3D coaxial reactor design

2. Reaction kinetics

- Two main reaction



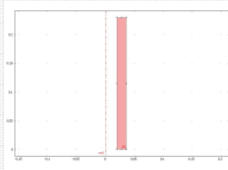
- Methanation of CO and CO₂ and RWGS reaction are negligible

- Rate equations determined by experimental investigations

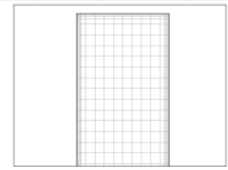
$$r_{\text{CO}} = 5 \cdot 10^{-5} \cdot \exp\left(\frac{-2.84 \cdot 10^4}{RT}\right) \cdot x_{\text{CO}}^{-0.5} \cdot x_{\text{O}_2} \quad x_i = \frac{c_i}{c_{\text{total}}}$$

$$r_{\text{H}_2} = 2.6 \cdot 10^{-7} \cdot \exp\left(\frac{-1.8 \cdot 10^4}{RT}\right) \cdot x_{\text{O}_2}^{1.16} \quad L = \frac{c_{\text{O}_2}}{2 \cdot c_{\text{CO}}}$$

3. Use of COMSOL Multiphysics



2D axial symmetric geometry



Rectangular mesh (mapped mesh)

Chemical engineering application modes:

Governing equations

Boundary conditions

- Mass transport (convection and diffusion)

$$\nabla \cdot (D_i \nabla c_i) = R_i - u \nabla c_i$$

$$R_i = -r_i \quad \text{bulk} \quad i = \text{H}_2, \text{CO} \quad R_{\text{O}_2} = -0.5 \cdot (r_{\text{CO}} + r_{\text{H}_2}) \quad \text{bulk}$$

- Energy balance (convection and conduction)

$$\nabla \cdot (-k \cdot \nabla T) = Q - \rho_{\text{gas}} \cdot C_p \cdot u \cdot \nabla T$$

$$Q = -\Delta H_{R,\text{CO}} \cdot R_{\text{CO}} - \Delta H_{R,\text{H}_2} \cdot R_{\text{H}_2}$$

- Momentum transport (Brinkman, porous media)

$$-\rho \cdot u = \nabla \left(p \cdot I + \frac{1}{3} \left((\nabla u + (\nabla u)^T) - \frac{2}{3} (\nabla u) I \right) \right)$$

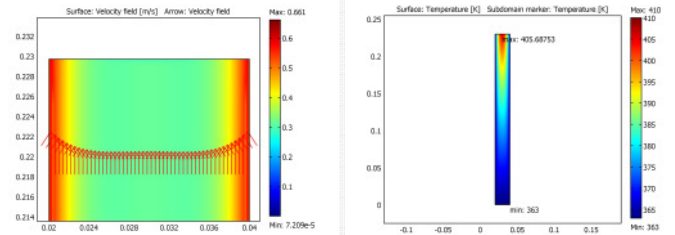
A porosity profile for is implemented

- Inlet → concentration
- Outlet → convective flux

- Inlet → T₀
- Outlet → convective flux
- Walls → heat flux
- q₀ = k_w · (T_w - T)

- Inlet → laminar flow
- Outlet → pressure

4. Results

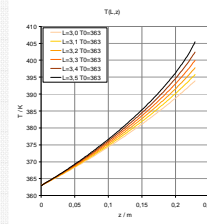


Radial velocity distribution as result of - profile

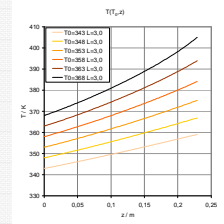
Exemplary surface plot of temperature

Cross section plots

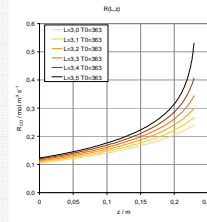
Lambda variation



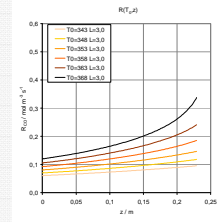
Inlet temperature variation



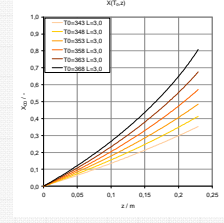
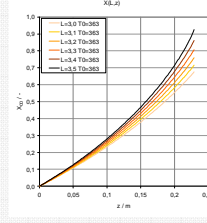
Reaction temperature T



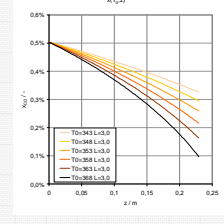
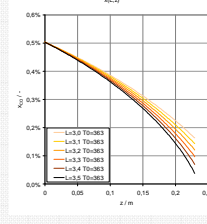
CO reaction rate R_{CO}



CO conversion rate X_{CO}



CO mole fraction x_{CO}



5. Conclusions

- A compact PROX reactor was designed.
- Two rate equations were determined within kinetic experiments.
- 2D geometry with axial symmetry was used in COMSOL Multiphysics v3.5a.
- Model combined mass, energy and momentum transport with chemical kinetics.
- Simulations confirmed model operability in principle and provided performance data.
- At higher reaction temperatures → no convergence or negative concentrations
- When the CO concentration approaches zero, numerical noise becomes significant in comparison to the concentration.
- Further CO conversion improvement is essential.