Presented at the COMSOL Conference 2009 Milan



Chemical Reactions in a µ-fluidic T-Sensor: Numerical Comparison of 2D and 3D Models

15 October 2009 | Remo Winz



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Why Cµ was formed:

> focus & merge expertise (chemistry, microsystems)

Cµ mission:

> explore chemical processes under develop of intelligent microsystem devices e.g. for lab-on-chip technology.

Lab-on-Chip Systems:

Combining fluidic parts with (electro-) analytical ones

Introduction

- Cµ & Lab-on-Chip
- T-Sensor

> 3D Model and 2D Projections

- Implementation
 - System Equations
 - Special requirements in 2D case

Numerical considerations in Comsol

- Comparison 2D / 3D
- Conclusion & Outlook

3D T-Sensor with Reaction

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Sensor Design

Projections from 3D to 2D

Projection P_1 :

Sensor Load:

 P_1

 P_2

The Mathematical Model

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Diffusion:	0.45e-9 m ² /s	Inlet Concentration:	1e-3 mol/m ³
Velocity:	6.6e-4 m/s	Receptors density:	3.32e-6 mol/m ²
Adsorption:	4.4e4 m ³ /mol/s		≈ 2 rec./nm²
Desorption:	1e-1 1/s	Degrees of Freedom:	~ 650.000 (P ₂)

Numerics & Implementation

Aim: Receive a stable and precise solution

Refined mesh in the region of strong kinetics

Numerical stabilization techniques

Direct vs. iterative Solvers

Solvers: PARDISO and GMRES (with ILU precond.) Computational Times: GMRES increasingly slow Accuracy: negligible influence

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fixed time steps
increasing effort

Accuracy

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compared with
'best' solution
no significant
improvement
major error at plug

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Mean Sensor Load 3D Simulation

14 July 2009

Adjusting k_f

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Comparison of Comsol and deal.II

Comsol Model
Streamline diffusion used

deal.II Model
uses Upwind scheme
fixed time stepping

Concentrations
 < 0.1 % of c_{max} dropped
Comparison of
 concentrations

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