

# Simulation of an Atmospheric Pressure Direct Current Microplasma Discharge in He/N<sub>2</sub>

Lizhu Tong<sup>1</sup>

<sup>1</sup>Kesoku Engineering System Co., Ltd.

1-9-5 Uchikanda, Chiyoda-ku, Tokyo 101-0047, Japan, tong@kesco.co.jp

**Abstract:** A study of an atmospheric pressure direct current microplasma discharge in He/N<sub>2</sub> is performed using COMSOL Multiphysics. The calculation of heat transfer is fully coupled with the plasma simulation so as to resolve the gas heating in discharges. A simple circuit model is used to decide the discharge voltage so that the current-voltage (*I-V*) characteristics are obtained. The *I-V* characteristics in He agree well with the available experimental data and it is found that with the addition of some amount of N<sub>2</sub> to He, N<sub>2</sub> is highly ionized. N<sub>2</sub> ions play an important role in the discharge kinetics, even if its amount is very small such as 0.002~0.02%.

**Keywords:** Microplasma, Atmospheric pressure, DC discharge, Gas heating.

## 1. Introduction

Microplasmas are characterized by their small size (characteristic dimensions, of tens to hundreds of microns) and high gas pressure (100 Torr–1 atm), yielding nonequilibrium plasmas[1]. Recently, atmospheric pressure nonequilibrium plasmas (APNEPs) have been widely studied for environmental and biological applications as well as material processing [2,3]. The atmospheric pressure direct current (dc) microdischarge, generally across a short gas gap between metal electrodes, is one of the easy methods of generating an APNEP.

For the extensive use of dc microplasmas at atmospheric pressure to meet industrial needs, it is essential to generate plasmas more efficiently and to understand the discharge properties, such as discharge current-voltage (*I-V*) characteristics, discharge gas composition, as well as the effect of gas temperature. The dc microdischarges at atmospheric pressure have been investigated in a He flow [1,4,5]. Wang and co-workers [1] used the fluid approach to simulate a dc atmospheric pressure microdischarge in He and they found that the microdischarge resembled a macroscopic low pressure dc discharge in many respects. They also indicated that gas heating played an

important role in atmospheric pressure microdischarges, in which the peak gas temperature increased from 370 to 650 K. Choi and co-workers [4] used the particle-in-cell Monte Carlo method to obtain the electron and ion kinetics in a dc atmospheric pressure microdischarge in He. A constant gas temperature of 300 K was used in their study. Belostotskiy and co-workers [5] reported their research about the effect of gas heating on the *I-V* characteristics in dc microdischarge in He at different operating pressures (300-800 Torr). Arkhipenko and co-workers [6] studied the effect of cathode temperature on atmospheric pressure dc glow discharges in He and they found that the cathode temperature varied from 350 to 900 K for the uncooled cathodes.

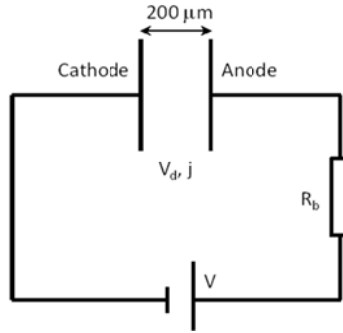
Some investigations of atmospheric pressure discharges in He/N<sub>2</sub> mixtures have also been carried out for analytical spectrometry [3,7,8]. The addition to He of some amount of a doping gas such as N<sub>2</sub> is shown to have a significant influence on the emission intensity of the analytes. Martens and co-workers [3] reported their research on atmospheric pressure glow discharge in He, including 10 ppm of nitrogen, by using fluid and Monte Carlo simulations. They indicated that at atmospheric pressure the positive column region became the dominant and that the He<sub>2</sub><sup>+</sup> ions played an important role. Wang and co-workers [7] performed their measurements by adding a small amount (0.02mol%) of N<sub>2</sub> to the He gas feed. The measured *I-V* characteristics were presented and the maximum gas temperature was found to be about 550 K. Petrov and co-workers [8] studied the capillary surface wave discharges at atmospheric pressure in He/N<sub>2</sub>. They indicated that the N<sub>2</sub> molecules were highly ionized through Penning ionization with the metastable He atoms.

However, the role of N<sub>2</sub> in He/N<sub>2</sub> microdischarges is still not completely clear, *e.g.*, the effect on the *I-V* characteristics. In this work, different small amounts of N<sub>2</sub> are added to the He gas to study the effect of gas composition on

dc microdischarge properties. The gas heating is calculated by solving the equation of heat transfer, coupled with the plasma simulation. The effect of cathode temperature on the  $I$ - $V$  characteristics is also studied.

## 2. Numerical Model

The model study for atmospheric pressure discharges is not similar to low-pressure discharges. It is known that the high pressure gives rise to more collisions and also to other types of collisions (such as three-body collisions, etc) [3]. In He/N<sub>2</sub> mixtures, a strong interaction between He and N<sub>2</sub> species takes place. The high rate coefficients of Penning ionization and of charge transfer reactions are expected to result in a decrease of He metastable atom density and in an increase of N<sub>2</sub> ion density.



**Figure 1.** Schematic of a microplasma model.

The schematic of a parallel-plate microplasma model is shown in Fig. 1. In this work,  $x = 0$  at cathode and  $x = 200 \mu\text{m}$  at anode. A dc power supplied voltage  $V$  and a ballast resistor  $R_b$  are used to generate a microdischarge.  $V_d$  is the discharge voltage and  $j$  is the discharge current density.  $V_d$  is solved by

$$V = V_d + jAR_b, \quad (1)$$

where  $A$  is the area of each electrode. In this work,  $A$  is set  $0.006 \text{ cm}^2$ , which is consistent with that used in the earlier experiment [7]. The obtained  $V_d$  is inputted into the plasma simulation as a boundary potential condition.  $j$  is obtained from the solution of the governing equations included transport of charged and neutral species, and Poisson's equation for the

electric field, which is performed in the Plasma Module in COMSOL Multiphysics. The species taken into account in the model include the ions  $\text{He}^+$ ,  $\text{He}_2^+$ ,  $\text{N}_2^+$ , the metastable He atoms:  $\text{He}(2^1\text{S})$ ,  $\text{He}(2^3\text{S})$ ,  $\text{He}^*$ , as well as electrons. Since only a small amount of N<sub>2</sub> is used, the excited N<sub>2</sub> molecules, the dissociated N atoms, and the ions  $\text{N}^+$ ,  $\text{N}_4^+$ ,  $\text{N}_3^+$ , are not considered in this work [2]. The reactions among species included in the model are given in Table 1, in which the cross sections  $\sigma$  and reaction rate coefficients  $k$  for electron impact collisions are taken from ref. [3,9,10]. The reaction rate coefficients  $k$  among heavy species (ions and neutral species) are taken from ref. [2,3,8].

**Table 1:** The reactions included in the model.

No.	Reaction	$\sigma/k$ ( $\text{cm}^3 \text{ s}^{-1} / \text{cm}^6 \text{ s}^{-1}$ )
1	$e^- + \text{He} \rightarrow e^- + \text{He}(2^1\text{S})$	$f(\varepsilon)$
2	$e^- + \text{He} \rightarrow e^- + \text{He}(2^3\text{S})$	$f(\varepsilon)$
3	$e^- + \text{He} \rightarrow 2e^- + \text{He}^+$	$f(\varepsilon)$
4	$e^- + \text{He}(2^1\text{S}) \rightarrow 2e^- + \text{He}^+$	$f(\varepsilon)$
5	$e^- + \text{He}(2^3\text{S}) \rightarrow 2e^- + \text{He}^+$	$f(\varepsilon)$
6	$2e^- + \text{He}^+ \rightarrow e^- + \text{He}$	$3 \times 10^{-20} (T_g/T_e)^4$
7	$2e^- + \text{He}^+ \rightarrow e^- + \text{He}^*$	$3 \times 10^{-20} (T_g/T_e)^4$
8	$e^- + \text{He}_2^+ \rightarrow \text{He} + \text{He}^*$	$8.9 \times 10^{-9} (T_g/T_e)^{1.5}$
9	$e^- + \text{N}_2 \rightarrow 2e^- + \text{N}_2^+$	$f(\varepsilon)$
10	$e^- + \text{N}_2^+ \rightarrow \text{N}_2$	$4.8 \times 10^{-7} (T_g/T_e)^{0.5}$
11	$\text{He}^+ + 2\text{He} \rightarrow \text{He}_2^+ + \text{He}$	$6.5 \times 10^{-32}$
12	$\text{He}(2^1\text{S}) + \text{He} \rightarrow 2\text{He} + h\nu$	$6.0 \times 10^{-15}$
13	$\text{He}(2^3\text{S}) + 2\text{He} \rightarrow \text{He}_2 + \text{He}$	$2.5 \times 10^{-34}$
14	$\text{He}(2^3\text{S}) + \text{He}(2^3\text{S}) \rightarrow \text{He}^+ + \text{He} + e^-$	$2.9 \times 10^{-9}$
15	$\text{He}^+ + \text{N}_2 \rightarrow \text{N}_2^+ + \text{He}$	$1.3 \times 10^{-9}$
16	$\text{He}_2^+ + \text{N}_2 \rightarrow \text{N}_2^+ + 2\text{He}$	$1.1 \times 10^{-9}$
17	$\text{He}^+ + \text{N}_2 + \text{He} \rightarrow \text{N}_2^+ + 2\text{He}$	$2.2 \times 10^{-29}$
18	$\text{He}_2^+ + \text{N}_2 + \text{He} \rightarrow \text{N}_2^+ + 3\text{He}$	$1.6 \times 10^{-29}$
19	$\text{He}(2^3\text{S}) + \text{N}_2 \rightarrow \text{N}_2^+ + \text{He} + e^-$	$6.3 \times 10^{-11}$
20	$\text{He}(2^1\text{S}) + \text{N}_2 \rightarrow \text{N}_2^+ + \text{He} + e^-$	$1.5 \times 10^{-10}$
21	$\text{He}(2^3\text{S}) + \text{N}_2 + \text{He} \rightarrow \text{N}_2^+ + 2\text{He} + e^-$	$2.9 \times 10^{-30}$

The heat transport of the gas is solved by

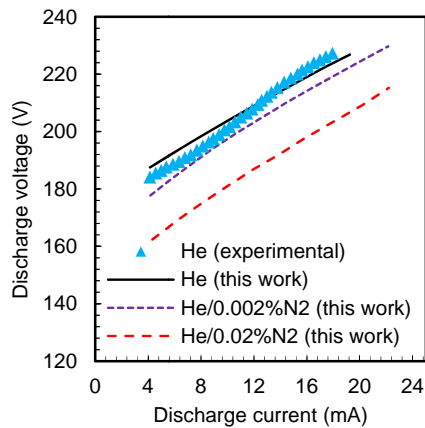
$$\rho c_p \frac{\partial T}{\partial t} = \nabla \cdot (k \nabla T) + Q, \quad (2)$$

where  $\rho$  is the gas density,  $T$  is the gas temperature,  $c_p$  is the specific heat,  $k$  is the thermal conductivity of the gas, and  $Q$  is the heat source term summing all the reactions shown in Table 1.

### 3. Results

#### 3.1 Current-voltage ( $I$ - $V$ ) characteristics

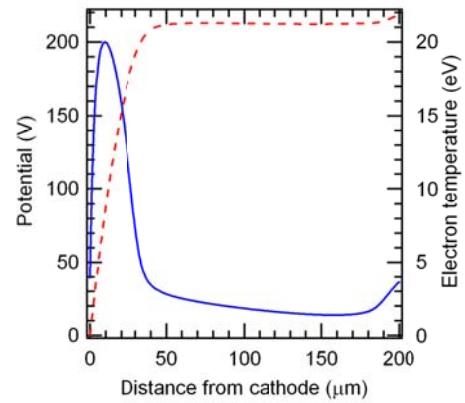
Figure 2 shows the simulated the  $I$ - $V$  characteristics. The boundary condition for electric field calculation is specified as  $V_c = 0$  on the cathode and  $V_a = V_d$  on the anode. The temperature for both electrodes is assumed to be 350 K. The secondary electron emission coefficient is decided by comparing the available experimental data [7]. It is found that at  $\gamma = 0.11$  for ions  $\text{He}^+$  and  $\text{He}_2^+$ , the calculated  $I$ - $V$  characteristics for pure He microdischarges are in good agreement with the experimental data. The same  $\gamma$  value is used for ions  $\text{N}_2^+$  in He/ $\text{N}_2$  microdischarges. The secondary electron temperature are assumed to be 5 eV [1]. The calculations are performed for pure He, He/0.002% $\text{N}_2$ , He/0.02% $\text{N}_2$  microdischarges. As seen in Fig. 2, even if a very small amount (0.002% mol)  $\text{N}_2$  is added, the  $I$ - $V$  characteristics are still affected. In the case of 0.02% mol  $\text{N}_2$  added in He, the  $I$ - $V$  characteristics greatly decrease.



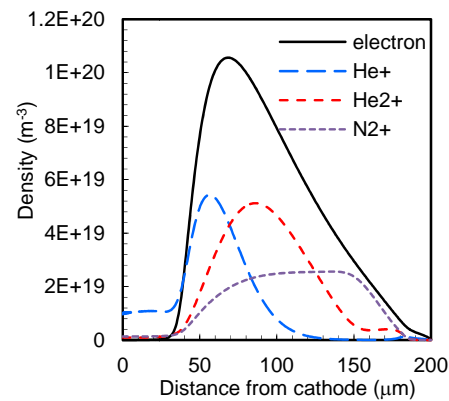
**Figure 2.** Current-voltage ( $I$ - $V$ ) characteristics for He, He/0.002% $\text{N}_2$ , and He/0.02% $\text{N}_2$  microdischarges.

#### 3.2 Discharge structure

The plasma properties for He/0.02% $\text{N}_2$  microdischarges at  $V = 420$  V are given in Figs. 3 and 4. As can be observed from Fig. 3, the electrical potential increases very quickly from 0 V at the cathode to about 210 V at a distance of 40  $\mu\text{m}$  from the cathode. This small region has been called as the cathode dark space (CDS) region [3]. Figure 3 also indicates a large positive column fills most of the discharge region. The electron temperature profile shows a maximum of 20 eV in the cathode sheath and decreases rapidly to much smaller values (about 1~2 eV). Figure 4 shows the density profiles of electrons and ions  $\text{He}^+$ ,  $\text{He}_2^+$ , and  $\text{N}_2^+$ . Although a small amount (0.02% mol) of  $\text{N}_2$  is added to He, the generated  $\text{N}_2^+$  ions are distributed over a large



**Figure 3.** Electrical potential (dashed line) and electron temperature (solid line) in He/0.02% $\text{N}_2$  microdischarges.



**Figure 4.** Density profiles for electrons and ions  $\text{He}^+$ ,  $\text{He}_2^+$ , and  $\text{N}_2^+$  in He/0.02% $\text{N}_2$  microdischarges.

discharge region with a maximum of about half of those of  $\text{He}^+$  and  $\text{He}_2^+$  ions, which could be deduced by the high Penning ionization of No. 19-21 and charge transfer reactions of No. 15-18 shown in Table 1.

### 3.3 Effect of cathode temperature

The simulated results for the effect of cathode temperature in  $\text{He}/0.02\%\text{N}_2$  microdischarges are given in Figs. 5-7. The cathode temperature is assumed to be 350, 450, and 550K and the anode temperature is fixed at 350 K. The other calculation conditions are the same as section 3.2. With the increase in cathode temperature from 350 to 550 K, the peak temperature of gas temperature increase from 580 to 700 K. It is well known that the electrical conductivity  $\sigma$  is strongly dependent on temperature, which can be approximated by

$$\sigma = \frac{1}{\rho_0[1+\alpha(T-T_0)]}, \quad (3)$$

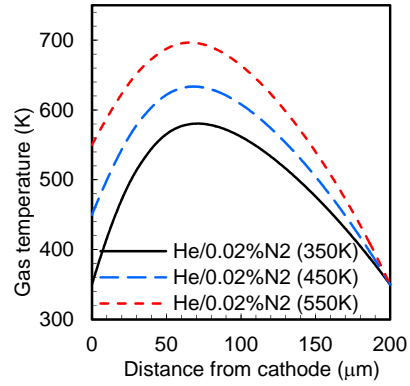
where  $\rho_0$  is the electrical resistivity at  $T_0$ ,  $\alpha$  is the proportional coefficient,  $T_0$  is the reference temperature,  $T$  is the gas temperature. The decrease of  $\sigma$  with the increase in temperature requests a higher discharge voltage to sustain the plasma discharge, as shown in Fig. 6.

Figure 7 shows the  $I$ - $V$  characteristics in  $\text{He}/0.02\%\text{N}_2$  microdischarges at different cathode temperatures. With the increase in cathode temperature, the  $I$ - $V$  characteristics rise up.

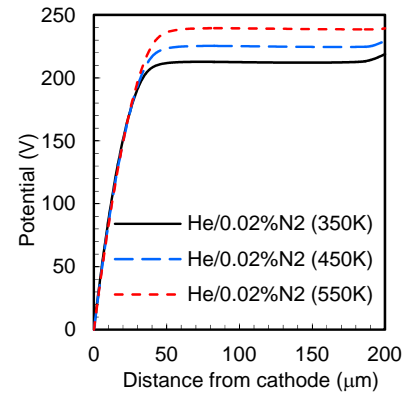
## 4. Conclusions

This paper presents the simulation results of atmospheric pressure direct current microplasma discharges in  $\text{He}/\text{N}_2$ . The fully coupled calculation of plasma discharge and heat transfer is realized. A simple circuit model is used to obtain the discharge voltage regarded as the boundary potential condition in the plasma simulation.

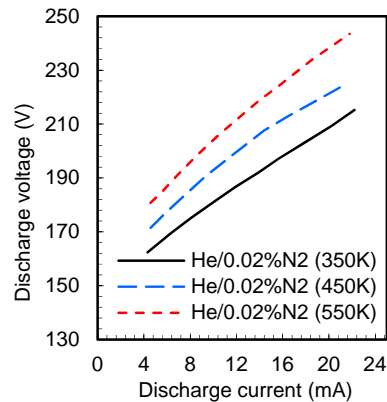
Results show that the present work efficiently reveals the plasma properties in  $\text{He}/\text{N}_2$  microdischarges, especially for the effect of a small amount of  $\text{N}_2$  added to  $\text{He}$  as well as the effect of cathode temperature on the  $I$ - $V$  characteristics. The present work can provide the useful information of the use of COMSOL



**Figure 5.** Gas temperature profiles in  $\text{He}/0.02\%\text{N}_2$  microdischarges at different cathode temperatures.



**Figure 6.** Electrical potential in  $\text{He}/0.02\%\text{N}_2$  microdischarges at different cathode temperatures.



**Figure 7.** Current-voltage ( $I$ - $V$ ) characteristics in  $\text{He}/0.02\%\text{N}_2$  microdischarges at different cathode temperatures.

Multiphysics in finding the design parameters of atmospheric pressure plasma sources for surface modification.

## 5. References

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